SYNTHESIS AND BIOLOGICAL ACTIVITY OF DERIVATIVES OF GLYCOPEPTIDE ANTIBIOTICS EREMOMYCIN AND VANCOMYCIN NITROSATED, ACYLATED OR CARBAMOYLATED AT THE *N*-TERMINAL

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Nitrosation, carbamoylation or acylation of the glycopeptide antibiotics eremomycin or vancomycin produced series of derivatives substituted at the N-terminus of the peptides. Though the modified amino group in these derivatives is not capable of protonation, N-nitroso derivatives retain antibacterial activity *in vitro* and *in vivo*. N-Carbamoyleremomycin has low activity, and N-Cbz-eremomycin and N-Boc-eremomycin are devoid of antibacterial activity, both *in vitro* and *in vivo*.

Eremomycin (I) is a potent glycopeptide antibiotic closely related to vancomycin¹⁾. Glycopeptide antibiotics are considered to act on bacterial cell wall biosynthesis at the level of transpeptidation/ transglycosylation by binding to peptides containing a terminal D-alanyl-D-alanyl residue, which are precursors to or the actual substrate for the transpeptidation reaction necessary for the formation of a rigid cell wall structure²⁾. The chemical modifications of eremomycin carried out in these studies were directed toward broadening the antibacterial spectrum and toward enhancement of chemotherapeutic properties³⁾. Elucidation of structure-activity relationships in these series is also of great importance. Recently it was shown that the affinity of eremomycin for the target site mimetic diacetyl-L-lysyl-D-alanyl-D-alanyl-D-alanine (DALAA) was 23-fold lower than that of vancomycin while the activity of eremomycin was greater than vancomycin against staphylococci and *Bacillus subtilis* and was less affected by DALAA than that of vancomycin⁴⁾. These data demonstrate a difference of eremomycin and indicate that structure-activity relationships found for vancomycin or teicoplanin may not be valid for eremomycin.

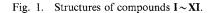
In this paper we study the possibility of modification of eremomycin and vancomycin with nitrosating, carbamoylating or acylating agents and the structure-activity relationships among *N*-substituted derivatives of these antibiotics (Fig. 1).

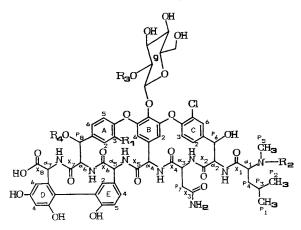
Chemistry

The nitrogen atom at the N-terminus of the peptide in eremomycin or vancomycin is designated as N, the nitrogen atom in the disaccharide branch of eremomycin (f fragment) or in vancomycin is designated as N', and the nitrogen atom in the monosaccharide branch of eremomycin (e fragment) is designated as N''.

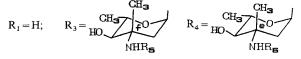
N-Nitrosated derivatives of eremomycin (II) (Scheme 1) and vancomycin (X) (Scheme 2) were prepared by the action of nitrous acid prepared *in situ* or by isoamylnitrite. The use of isoamylnitrite permitted the study of the nitrosation reaction in DMSO at pH $2 \sim 10$. Both I and IX are nitrosated only at pH 3 the major reaction products (II or X respectively) being N–NO derivatives. The direction of electrophilic attack seems to be rather unexpected as in the molecules of I or IX there are three other groups of nucleophilic centers susceptible to nitrosation, *i.e.* amino groups of sugars, hydroxylated aromatic rings and an asparagine amido group. In our experimental conditions neither of these centers in the cremomycin or vancomycin molecule were affected, and this may be explained by the tertiary structure of the antibiotic in which some parts of the molecule are more and other less hindered. This may lead to the decreased of reactivity of some groups and enhancement of others.

Reductive alkylation of II or X with benzaldehyde and NaBH₃CN gave N', N''-dibenzyl-*N*nitrosoeremomycin (III) or N'-benzyl-*N*-nitrosovancomycin (XI), respectively. Interaction of II with





Eremomycin and its derivatives

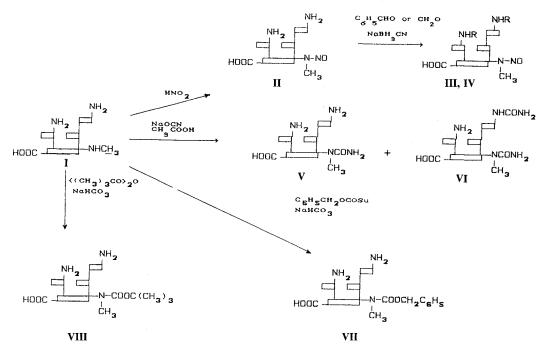


- $I R_2 = R_5 = R_6 = H$
- II $R_2 = NO; R_5 = R_6 = H$
- III $R_2 = NO; R_5 = R_6 = CH_2C_6H_5$
- IV $R_2 = NO; R_5 = R_6 = CH_3$
- $V = R_2 = CONH_2; R_5 = R_6 = H$
- **VI** $R_2 = R_5 = CONH_2; R_6 = H$
- **VII** $R_2 = COOCH_2C_6H_5; R_5 = R_6 = H$
- **VIII** $R_2 = COOC(CH_3)_3; R_5 = R_6 = H$

Vancomycin and its derivatives

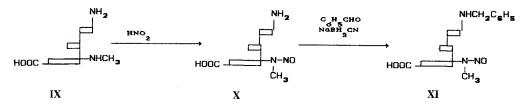
$$R_1 = Cl;$$
 $R_4 = H;$ $R_3 = \begin{array}{c} CH_3 \\ \hline FO \\ HO \end{array}$

- $IX \qquad R_2 = R_7 = H$
- $\mathbf{X} = \mathbf{R}_2 = \mathbf{NO}; \ \mathbf{R}_7 = \mathbf{H}$
- **XI** $R_2 = NO; R_7 = CH_2C_6H_5$



Scheme 1. Chemical transformations of eremomycin.

Scheme 2. Chemical transformations of vancomycin.



formaldehyde and reduction with NaBH₃CN gave N', N''-dimethyl-N-nitrosoeremomycin (IV).

N-Cbz-eremomycin (VII) and *N*-Boc-eremomycin (VIII) were obtained by the interaction of I with *N*-(benzyloxycarbonyloxy)succinimide (CbzOSu) or di-*tert*-butyl dicarbonate (Boc₂O) in water-dioxane mixture (1:1) in the presence of NaHCO₃.

N-Carbamoyl-eremomycin (V) and N,N'-dicarbamoyleremomycin (VI) were obtained by the interaction of NaNCO and I in the presence of acetic acid in DMSO, the directions of substitution being similar to those in the previously described acylation reaction of I⁵).

Isolation and purification of compounds II, IV, V, VI, VIII, and X were performed by means of ion exchange chromatography on CM-cellulose similar to purification of N-acyleremomycins⁵⁾ and eremomycin esters⁶⁾.

Functional group electrophoretic and TLC analyses of these new derivatives are summarized in Table 1. Positions of the introduced nitroso, carbamoyl, acyl and alkyl groups were determined after acidic hydrolysis and Edman degradation. Previously it was demonstrate that hydrolysis in $0.2 \times \text{HC}$ at 100°C of eremomycin during 10 minutes leads to the splitting of the glycosidic bond between the two sugars in the disaccharide branch and gives rise to eremosamine⁵). TLC comparisons of the hydrolysis products

	Acidic hydroly	vsis presence of	Edman degradation	¹ H NMR		TLC R	f values
Com-	eremo	samine	presence of	number of introduced	Electrophoretic mobility relative	Syst	ems
pound	in disaccharide branch	in monosac- charide branch	<i>N</i> -terminal <i>N</i> -methylleucine	alkyl or acyl groups	to eremomycin [–]	A	В
I	+	+	+	_	1.00	0.30	0.05
II	+	+	_		0.75	0.45	0.45
III		_	_	2	0.70	0.55	0.52
IV	_	_	_	2	0.72	0.48	0.48
V	+	+	-		0.76	0.26	0.20
VI	-	+	_		0.55	0.29	0.50
VII	+	+	_	1	0.74	0.50	0.56
VIII			_	1	0.73	0.46	0.54
	Presence of	vancoseamine					
	In disaccharide branch	Sugar is absent	_				
IX	+		+		0.75	0.40	0.07
Х	+		_		0.54	0.52	0.49
XI	_		-	1	0.50	0.60	0.57

Table 1. Results of analytical investigation of compounds $I \sim XI$.

of compounds III, IV, VI with an authentic sample of eremosamine demonstrated that in these compounds the aminosugar formed is different from eremosamine. Hydolysis of XI led to an aminosugar different from vancosamine. Eremosamine is one of the products of hydroxysis of II, V or VII. Hydrolysis of XI gave vancosamine. TLC comparisons of the more drastic hydrolysis products of compounds III and IV (1N HCl, 100°C, 30 minutes), when glucose as well as eremosamine from the monosachharide branch are split off, with an authentic sample of eremosamine demonstrated that in these compounds the aminosugars formed are different from eremosamine. Among the drastic hydrolysis products of compounds II, V, VI, VII eremosamine was present.

Hydrolysis of VIII removed the Boc-group.

Edman degradation⁷) showed that in all the derivatives the *N*-methylamino group is substituted: TLC demonstrated the adsence of 5-isopropyl-1-methyl-3-phenyl-2-thiohydantoin by the comparison with an authentic sample of the latter, which was obtained in a parallel experiment by Edman degradation of eremomycin.

Electrophoresis in $2 \times CH_3COOH$ demonstrated the decreased number of positively charged amino groups in compounds II, V, VII and X in comparison with the parent antibiotic and showed that the number of these groups in III, IV, XI is similar to the starting II or X, respectively. Compound VI is less positively changed than V.

The ¹H NMR spectra of the new compounds contained all the ¹H signals of eremomycin (compounds $I \sim VIII$) or vancomycin (compounds X, XI), backbone as well as those of the introduced groups (See Experimental).

The ¹³C NMR spectra of compounds II, V and X are presented in Table 2. ¹³C NMR signals were assigned by comparison with those of eremomycin and vancomycin⁸⁾.

Treatment of I with Na¹⁵NO₂ in the presence of acetic acid gave ¹⁵NO-eremomycin (¹⁵N-II). The ¹⁵N NMR spectrum showed two signals at 513.0 and 509.0 ppm (\sim 3:1) in the area of *N*-nitrosoamino

Table 2. ¹³C NMR chemical shifts (δ_c , ppm) of eremomycin (I), vancomycin (IX) and their derivatives II, V and X.

Assignment	I ^a	¹⁵ N-II ^b	V°	IX ^a	Хь
	(D ₂ O)	$(D_2O + DMSO)$	$(D_2O + DMSO)$	(D ₂ O)	$(D_2O + DMSO)$
x-8	176.40	176.98	176.74	177.30	$\sim 177^{d}$
x-3	174.46	174.34	175.12	175.10	175.82
x-5	172.11	172.34	173.41	171.20	172.14
x-6	171.80	172.30	172.04	169.80	170.88
x-4	171.29	172.06	171.40	169.00	169.44
x-1	171.23	171.59	170.77	168.00	169.01
x-2	169.39	170.25	169.38	171.90	173.80
x-7	167.88	168.21	167.42	170.20	171.26
D-3	157.21	157.72	~157 ^d	157.30	157.67
B-3	157.21	157.68	~157 ^d	152.20	153.10
A-4	156.41	157.01 ^d	156.83	149.80	149.85
D-5	155.68	156.47	155.39	149.80	149.85
E-6	155.08	155.70	155.59	155.00	155.64
			~153 ^d		
B-5	153.34	153.6 ^d		153.60	153.35
C-4	150.76	151.06	150.27	151.40	151.46
C-1	138.20	139.23	138.37	139.50	139.40
D-1	137.92	138.86	137.87	138.50	138.75
E-2	136.31	136.98	136.25	136.50	136.75
B-1	134.92	135.65	136.22	135.90	134.74
B-4	134.22	135.58	134.83	133.40	133.44
A-1	133.83	134.6 ^d	∼134 ^ª	141.20	142.11
C-6	130.70	130.79	129.69	129.60	130.18
A-2	129.42	130.03	129.06	129.10	129.10
A-6	128.79	129.47	128.89	128.70	128.84
C-5	128.35	129.15	128.70	127.50	127.45
E-4	127.60	127.79	127.24	127.30	127.24
C-2	127.28	127.18	126.10	128.10	128.33
E-3	126.56	127.12	125.91	128.10	128.33
C-3	125.56	125.73	124.86	125.30	125.18
A-3	123.55	123.83	123.18	124.70	124.53
E-1	122.17	123.06	121.91	122.20	122.83
A-5	122.09	122.75	121.81	127.10	127.08
D-2	118.27	119.12	118.15	118.50	118.56
D-6	109.09	109.33	108.65	109.10	108.61
B-2	107.71	107.9 ^d	108.65	109.00	107.86
B-2 B-6	107.71	107.9	104.01	105.00	107.80
Б-0 D-4		105.07	103.35	103.90	103.83
	103.87 102.30	$\sim 103^{d}$	103.29	103.30	102.45
g-1					
f-1	98.03	98.41	97.70	98.50	98.29
e-1	93.27	93.54	92.87		
g-2	80.00	80.3 ^d	$\sim 79^{d}$	80.10	79.75
g-3	77.14	77.4 ^d	$\sim 77^{d}$	76.90	77.23
g-5	76.44	76.13	$\sim 76.5^{d}$	77.50	77.61
e-4	75.62	75.78	75.42		
P-8	75.30	75.42	75.15	72.40	71.76
f-4	75.20	75.42	74.73	71.80	71.70
P-6	71.78	71.66	70.48	72.90	72.26
g-4	70.33	70.80	69.93	70.30	70.76
e-5	67.03	67.38	66.66	_	—
f-5	66.62	67.09	66.34	64.90	65.48
α-6	62.31	63.52	63.10	64.10	64.50
α-1	61.76	62.39	61.48	61.70	63.70
g-6	61.76	60.12	59.43	61.60	62.08
α-2	59.83	60.06	59.09	59.60	59.52

Assignment	I ^a (D ₂ O)	15 N-II ^b (D ₂ O + DMSO)	V^{c} (D ₂ O+DMSO)	IX ^a (D ₂ O)	X^{b} (D ₂ O+DMSO)
α-7	59.46	60.06	58.50	60.40	59.90
e-3	57.45	57.84	57.14		_
f-3	57.11	57.43	56.66	55.60	55.46
α-5	55.34	55.87	55.14	54.90	55.13
α-4	55.21	55.57	54.88	56.00	56.18
α-3	53.10	53.44	52.61	52.50	52.30
P-4	39.76	40.14	39.37	39.60	40.38
f-2	39.40	39.56	39.70	34.00	34.00
e-2	39.08	38.37	38.25		
P- 7	36.95	37.12	36.25	36.50	37.16
P-5	32.66	31.13	31.72	32.90	30.77
P-3	24.53	25.41	24.86	24.90	25.30
P-1	22.78	23.56	22.64	23.10	23.46
P-2	22.03	22.00	21.12	22.80	22.02
Me-e-3	18.85	19.42	18.54	·	_
Me-f-3	18.51	19.16	18.21	22.90	23.19
Me-e-5	17.91	18.65	17.68		·
Me-f-5	17.59	18.30	$\sim 17^{d}$	17.10	17.41

Table 2. (Continued)

^a Data from⁸⁾.

^b In the high field part of the spectra signals of the minor isomer of ¹⁵N-II at 35.93, 31.53, 25.64, 23.41, 22.46 ppm are observed; signals of the minor isomer of **X** are at 36.10, 25.57, 23.27, 22.50 ppm.

[°] Signal of CONH₂ is observed at 162.08 ppm.

^d Signals are broad.

compounds⁹⁾. ¹⁵N signals appeared to correspond to the *syn* and *anti* isomers of **II**. In the ¹³C NMR spectra of **II** and **X** signals of minor isomers can be also seen (Table 2).

Results and Discussion

Table 3 shows the *in vitro* antibacterial activity of all synthesized compounds against *B. subtilis* and 25 clinical isolates of methicillin-resistant *Staphylococcus aureus* in comparison with the parent antibiotics I and IX. The activity of compounds II, III, X and XI against *B. subtilis* is comparable to that of vancomycin and is two fold lower than the activity of eremomycin. Compounds IV, V, VI, VII, VIII have much lower activity against *B. subtilis*.

C	B. substilis MIC (μg/ml)	Methicillin-resistant S. aureus (N=25)		
Compound		MIC ₅₀ (μg/ml)	MIC ₉₀ (μg/ml)	
I	0.08	0.5	1	
II	0.16	2	4	
Ш	0.16	4	8	
IV	1.60	16	32	
V	3.20	16	32	
VI	4.00	> 32	> 32	
VII	4.00	> 32	> 32	
VIII	4.00	> 32	> 32	
IX	0.16	1	2	
Х	0.16	4	8	
XI	0.16	2	4	

Table 3. In vitro activity of eremomycin (I), vancomycin (IX) and their derivatives (II \sim VIII, X, XI).

Against S. aureus, compounds II, III, X, and XI are $4 \sim 8$ times less active than eremomycin and $2 \sim 4$ times less active than vancomycin. Compounds IV, V have poor activity against S. aureus while compounds VI, VII and VIII are devoid of activity.

A similar response was noted when the nitroso compounds II and X were studied against greater numbers of Gram-positive clinical isolates (Table 4). Eremomycin (I) inhibited all Gram-positive isolates at a concentration $<0.50 \,\mu$ g/ml, vancomycin (IX) at a concentration $<2 \,\mu$ g/ml, nitroso derivatives II and Organism

(No. tested)

S. aureus MR (8)

S. aureus MS (8)

Coagulase negative

Coagulase negative

staphylococci MR (7)

staphylococci MS (7)

Enterococcus faecium PT (9)

Enterococcus faecalis PT (10)

MIC (µg/ml)				
Range	MIC ₅₀	MIC ₉₀		
0.25~ 0.50	0.25	0.50		
1.00~ 8.00	2.00	4.00		
0.50~ 2.00	1.00	2.00		
4.00~16.00	4.00	8.00		

0.25

2.00

0.50

4.00

0.50

4.00

2.00

8.00

0.25

2.00

0.50

8.00

0.25

1.00

0.50

4.00

0.50

4.00

2.00

16.00

0.25

1.00

0.50

4.00

0.25

2.00

1.00

8.00

0.12

1.00

0.50

4.00

0.06

0.50

0.25

4.00

0.25

1.00

0.25

4.00

Table 4. In vitro activity of eremomycin (I), vancomycin (IX), N-nitrosoeremomycin (II) and N-nitrosovancomycin (X) against gram-positive clinical isolates.

0.12~ 0.25

1.00~ 2.00

 $0.12 \sim 0.50$

 $2.00 \sim 8.00$

1.00~ 2.00

 $4.00 \sim 16.00$

0.12~ 0.25

1.00~ 2.00

0.50~ 1.00

 $4.00 \sim 8.00$

 $0.03 \sim 0.25$

0.12~ 1.00

0.12~ 0.50

1.00~ 4.00

0.03~ 0.50

 $0.50 \sim 4.00$

 $0.12 \sim 2.00$

 $1.00 \sim 16.00$

0.50~

4.00~

Compound

I

II

IX

Х

I

II

IX

Х

I

II

IX

Х

I

П

IX

Х

Ĩ

II

IX

Х

I

Π

IX

Х

X were at least 8 fold less active (at a concentration $< 8 \mu g/ml$ for **II** and at a concentration $< 16 \mu g/ml$ for **X**). Compounds **II** and **X** like the parent antibiotics **I** and **IX** were not active against *Escherichia coli* (data not shown).

Comparison of the therapeutic effects of eremomycin and its nitroso derivative II against systemic staphylococcal infection in mice demonstrated that II was only three fold less active than I: ED_{50} for I is 2.50 mg/kg; ED_{50} for II is 7.50 mg/kg. Compounds V, VII and VIII were far less active than I.

Experimental

Eremomycin was produced at the Pilot Plant of the Institute of New Antibiotics, Russian Academy of Medical Sciences. All other reagents and solvents used were commercial products. ¹⁵N NaNO₂ (95% isotopic enrichment) was obtained from ISOTOP (USSR). All NMR measurements were obtained with a Varian VXR-400 instrument operated at 400 MHz for ¹H, at 100.6 MHz for ¹³C and at 40.5 MHz for ¹⁵N. ¹H NMR spectra were recorded at 80°C, ¹³C NMR spectra were obtained at 70°C using signals of solvent (DMSO- d_6) as an internal standard (δ 39.5 ppm). ¹⁵N NMR spectra were recorded at 70°C with gated decoupling. ¹⁵N chemical shifts were measured from NH₄ (δ 0 ppm) as an external standard.

CM-cellulose column chromatography was performed on a LKB Ultragrad Gradient Mixer 11300, supplied with Uvicord 2138 and Recorder 2065. Precoated Merck Silica gel $60F_{254}$ plates were used for TLC in systems: ProOH - EtOAc - conc. NH₄OH, 3:3:4 (A) and BuOH - AcOH - H₂O, 5:2:3 (B). Paper electrophoresis was carried out in 2 N AcOH at 700 V for 3 hours. Ion-exchange resin SDW-3, an analog of Dowex 50X2, was obtained from BioLar, Olaine, Latvia.

N-Nitrosoeremomycin (II)

A. Isoamyl nitrite (0.067 ml, 0.5 mmol) and 0.5 ml of AcOH were added to a stirred solution of

eremomycin sulfate (330 mg, 0.2 mmol) in 15 ml of DMSO. The reaction mixture was stirred at room temperature for 5 hours. After addition of 200 ml of acetone the precipitated solid was collected by filtration and purified by CM-cellulose column chromatography (column 1×40 cm) on CM-32 (Whatman) using 0.02 M AcONH₄ - 0.2 M AcONH₄ gradient (8 hours). The solution was adjusted to pH 2, passed through ion exchange column (0.5×6 cm) of SDW-3 (H⁺) and eluted with $0.25 \times \text{NH}_4$ OH. After concentration of the eluate to 3 ml and neutralization to pH 6 with $6 \times \text{H}_2$ SO₄ 100 ml of acetone was added to give a precipitate of II (210 mg, 61%).

B. NaNO₂ (14 mg, 0.2 mmol) and 0.1 ml of AcOH were added to a stirred solution of eremomycin sulfate (330 mg, 0.2 mmol) in 10 ml of H_2O -DMSO (4:1). The reaction mixture was stirred at room temperature for 0.5 hour. Isolation and purification of **II** was performed as described above.

N-Nitrosovancomycin (X)

A. X (185 mg, 61%) was obtained from 297 mg (0.2 mmol) of vancomycin hydrochloride and 0.067 ml (0.5 mmol) of isoamyl nitrite by a method similar to that described for **II**.

B. X (145 mg, 48%) was obtained from vancomycin hydrochloride (297 mg, 0.2 mmol) and 14 mg (0.2 mmol) of NaNO₂ by the method similar to that described for II.

N', N''-Dibenzyl-N-nitrosoeremomycin (III)

To a stirred solution of II (168 mg, 0.1 mmol) in 8 ml of DMSO, C_6H_5CHO (0.061 ml, 0.6 mmol) and NaBH₃CN (25 mg, 0.4 mmol) were added. The reaction mixture was stirred at room temperature for 6 hours and a product was precipitated with 150 ml of Et₂O. It was collected by filtration and redissolved in 90 ml of a mixture of BuOH - H₂O (2:1). The resulting solution was washed with H₂O (60 ml × 2), the organic layer was separated and concentrated to a small volume (~2 ml). After addition of 100 ml of Et₂O the precipitate was collected by filtration to yield 140 mg (75%) of III.

¹H NMR (CD₃OD): δ 7.3 ~ 7.6 ppm (10H, m, 2 × C₆H₅).

N'-Benzyl-*N*-nitrosovancomycin (XI)

Starting from X (151 mg, 0.1 mmol) and using substantially the same alkylation procedure described for III 125 mg (78%) of the pure title compound was obtained.

¹H NMR (CD₃OD): δ 7.3 ~ 7.6 ppm (5H, m, C₆H₅).

N', N''-Dimethyl-N-nitrosoeremomycin (IV)

To a stirred solution of II (168 mg, 0.1 mmol) in 8 ml of a mixture of H_2O -DMSO (1:1), a 37% (w/v) water solution of formaldehyde (0.075 ml, 1 mmol) and 50 mg (0.8 mmol) of NaBH₃CN were added. The reaction mixture was stirred at room temperature for 4 hours and was precipitated with 100 ml of acetone. The purification was performed as described for II to give 90 mg (53%) of IV.

¹H NMR (D₂O - DMSO, 2:1): δ 3.0 ppm (6H, s, CH₃ × 2).

N-Carbamoyleremomycin (V) and N, N''-dicarbamoyleremomycin (VI)

NaNCO (20 mg, 0.3 mmol) and AcOH (0.6 ml) were added to a stirred solution of eremomycin sulfate (330 mg, 0.2 mmol) in 15 ml of DMSO. After 15 hours 150 ml of acetone was added to give a precipitate of a mixture of V and VI, which were separated and purified by the method described for II to yield 160 mg (47%) of V and 80 mg (23%) of VI.

N-Carbobenzoxyeremomycin (VII)

CbzOSu (100 ml, 0.4 mmol) and NaHCO₃ (17 mg, 0.2 mmol) were added to a stirred solution of eremomycin sulfate (330 mg, 0.2 mmol) in 10 ml of H₂O-dioxane mixture (1:1) at room temperature. After 10 hours 150 ml of acetone was added. Purification of the precipitate was performed as described for III to give 150 mg (42%) of VII.

¹H NMR (D₂O - DMSO, 2:1): δ 7.3 ~ 7.6 ppm (5H, m, C₆H₅).

N-Boc-eremomycin (VIII)

Boc₂O (109 mg, 0.5 mmol) and NaHCO₃ (17 mg, 0.2 mmol) were added to a stirred solution of

¹H NMR (D_2O - DMSO, 2:1): δ 0.9 ppm (9H, s, CMe₃).

In vitro antibacterial activity against *Bacillus subtilis* ATCC 6633 was determined by the microdilution method in tubes in medium containing: 1,000 ml of Hottinger hydrolysate with 28 mg% of amine N, 30 g of agar, 5 g of peptone and 2.5 g of CaCO₃. Sterilization was performed at pH $7.2 \sim 7.4$. Activity of the compounds against 25 methicillin resistant clinical isolates of *Staphylococcus aureus* was determined by the agar dilution method. Agar plates containing 15 ml of agar containing two-fold serial dilutions of the test agents were inoculated to yield a final density of approximetely 10^7 CFU per spot. The plates were incubated at 35°C for 24 hours in ambient air.

Minimal inhibitory concentrations (MIC's) of I, II, IX, and X against an expanded spectrum of Gram-positive bacteria (Table 4) were determined by the microdilution method as recommended by the National Committee for Clinical Laboratory Standards¹⁰. The tests were performed in 96 well microtiter plates using Mueller-Hinton II broth with a final volume of 0.1 ml per well and an inoculum density of approximately 10^5 CFU/ml. The medium was supplemented with 5% sheep blood cells when testing non enterococcal streptococci. Plates were incubated at 35° C for $18 \sim 24$ hours in ambient air. MIC values were defined as the lowest concentration of antimicrobial agent that completely inhibits growth of the organism as detected by the unaided eye.

Antibacterial activity in vivo. The therapeutical effect of I and II were determined against acute lethal staphylococcal infection in albino mice. Mice (strain $BDF_1 (20 \pm 2 g)$) were challenged by intraperitoneal injections with sufficient bacterial cells suspended in hog gastric mucin to kill nontreated control within 24~48 hours. The test compounds were administered subcutaneously 30 minutes and 24 hours post infection. In each test 5 mice were treated per one dose level. On the 7th day ED_{50} were calculated on the basis of the percentage of surviving mice at each dose using Berens method.

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